# Structure of 2-Amino-5-bromopyrimidine 

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#### Abstract

C}_{4} \mathrm{H}_{4} \mathrm{BrN}_{3}, \quad M_{r}=174.01\), orthorhombic, Cтса,$\quad a=7.702$ (3),$\quad b=8.215$ (3), $\quad c=$ 17.733 (12) $\AA, \quad U=1122.0 \AA^{3}, \quad Z=8, \quad D_{x}=$ $2.06 \mathrm{~g} \mathrm{~cm}^{-3}$, Mo $K \alpha$ radiation, $\lambda=0.71069 \AA, \mu=$ $71.4 \mathrm{~cm}^{-1}, F(000)=672, T=293 \mathrm{~K}, R=0.060$ for 578 unique observed $[F>5 \sigma(F)]$ reflections. Atoms Br , $\mathrm{C} 5, \mathrm{C} 2$ and N 2 lie on the mirror plane at $x=0$; atoms N 3 and C 4 are therefore symmetry equivalents of N 1 and C6. The numbering system conforms to that for pyrimidine nucleobases. The molecules of the title compound form self-base-paired ribbons which run parallel to the $a$ axis. The hydrogen bond involved in the base-pairing between symmetry related molecules is $\mathrm{N} 2 \cdots \mathrm{~N} 1(-0.5-x, 1 \cdot 5-y,-z) 3 \cdot 10$ (3) $\AA$.


Experimental. Crystals were obtained from aqueous solution as very thin micaceous plates. Space group and initial cell dimensions were obtained from Weissenberg photographs. Data were collected on a Nicolet P3 (four-circle) diffractometer in Aberdeen by RAH. The crystal had dimensions $0.04 \times 0.4 \times 0.6 \mathrm{~mm}$. Cell parameters were measured on the diffractometer using 14 reflections in the $2 \theta$ range $17-21^{\circ}$. Range of indices: $0 \leq h \leq 12 ; 0 \leq k \leq 12 ; 0 \leq l \leq 24$. Data measured using $\omega / 2 \theta$ scans in the range $0<2 \theta<60^{\circ}$. Standard reflections, 020 and 120 , were measured every 50 reflections. No changes greater than $2 \sigma$ from the mean of the intensities of these reflections were found throughout data collection. 783 independent reflections measured, giving 578 observed $[F>5 \sigma(F)]$ reflections

Table 1. Coordinates $\left(\times 10^{4}\right)$ and $U_{\text {eq }}$ values $\left(\AA^{2} \times 10^{3}\right)$ for non-hydrogen atoms with e.s.d.'s in parentheses

| $U_{\text {eq }}=\frac{1}{3} \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eo }}$ |
| Br | 0 | 2636 (1) | 2251 (1) | 46 (1) |
| N1 | -1557 (6) | 6200 (5) | 773 (2) | 35 (1) |
| C2 | 0 | 6755 (9) | 548 (4) | 32 (2) |
| N2 | 0 | 8003 (9) | 48 (5) | 42 (1) |
| C6 | -1541 (8) | 4997 (6) | 1283 (3) | 35 (1) |
| C5 | 0 | 4362 (9) | 1539 (4) | 35 (2) |

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used in the refinement. Lorentz and polarization factors were applied. A $\psi$-scan absorption correction was used. Maximum and minimum transmission factors were 0.3748 and 0.0567 . The structure was solved by the heavy-atom method using the SHELXS86 program (Sheldrick, 1986).

Blocked full-matrix refinement (on $F$ ) was carried out using the program SHELX76 (Sheldrick, 1976). The hydrogen atoms were included at calculated positions. All hydrogen atoms were given isotropic temperature factors 1.5 times that of the parent atom and allowed to ride on that atom. All other atoms were refined anisotropically. The refinement converged at $R=$ $0.060, w R=0.071, w=2.5097\left[\sigma^{2}(F)+0.001511 F^{2}\right]^{-1}$. 51 refined parameters; max. shift/e.s.d. $<0.01$; max. difference peak 1.29 , min. difference peak $-1.36 \mathrm{e} \AA^{-3}$. The largest of these difference peaks were associated with the Br atom.

Scattering factors and anomalous-dispersion corrections were taken from International Tables for X-ray Crystallography (1974). Also used were the program packages XANADU (Roberts \& Sheldrick, 1975) and PLUTO (Motherwell \& Clegg, 1978). All calculations were carried out on the Dundee University Prime


Fig. 1. Perspective view of the molecule showing atomic numbering (see Abstract).
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Table 2. Interatomic distances $(\AA)$ and angles $\left({ }^{\circ}\right)$

| $\mathrm{C} 5-\mathrm{Br}$ | $1.899(7)$ | $\mathrm{C} 2-\mathrm{N} 1$ | $1.344(5)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{C} 6-\mathrm{N} 1$ | $1.339(6)$ | $\mathrm{N} 2-\mathrm{C} 2$ | $1.355(12)$ |
| $\mathrm{C} 5-\mathrm{C} 6$ | $1.374(7)$ |  |  |
| $\mathrm{C} 6-\mathrm{N} 1-\mathrm{C} 2$ |  | $116.3(5)$ | $\mathrm{N} 1-\mathrm{C} 2-\mathrm{N} 1$ |
| $\mathrm{~N} 2-\mathrm{C} 2-\mathrm{N} 1$ | $116.8(4)$ | $\mathrm{C} 5-\mathrm{C} 6-\mathrm{N} 1$ | $126.3(7)$ |
| $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 6$ | $119.5(7)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{Br}$ | $120.8(6)$ |

1 and 2). Fig. 2 shows a view of the molecular packing.*

Related literature. Base-pairing between nucleosides, nucleotides and nucleobases has been reviewed by Wilson \& Tollin (1987).

[^0]* Lists of structure factors, anisotropic thermal parameters and
H-atom parameters have been deposited with the British Library
Document Supply Centre as Supplementary Publication No. SUP
51167 (7 pp.). Copies may be obtained through The Executive
Secretary, International Union of Crystallography, 5 Abbey
Square, Chester CH1 2HU, England.


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computer. The atomic numbering is shown in the perspective drawing (Fig. 1), and tables of atomic parameters, bond lengths and angles are given (Tables

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# Structure of Intrasil Brilliant Yellow* 

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#### Abstract

Chloro-2-benzoxazolyl)-7-(diethyl-amino)-2 H -1-benzopyran-2-one, $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{ClN}_{2} \mathrm{O}_{3}, M_{r}=$ 369, monoclinic, $P 2_{1} / a, a=9.262$ (2), $b=13.282$ (2), $c=14.453$ (2) $\AA, \quad \beta=104 \cdot 16(2)^{\circ}, \quad V=1724(1) \AA^{3}$, $Z=4, D_{m}$ (flotation in KI solution) $=1.43, D_{x}=$ $1.42 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{Mo} K \alpha)=0.7107 \AA, \quad \mu=2.51 \mathrm{~cm}^{-1}$, $F(000)=768, T=293 \mathrm{~K}, R=0.056$ for 1071 observed reflections. The dihedral angle between the benzoxazoline and the benzopyrone moieties is $5.9^{\circ}$, showing significant deviation from planarity. The two ring systems are planar.

Experimental. The title compound is a commercial dye. It was dissolved in chloroform and purified by column


[^1]chromatography using $20 \%$ chloroform and $80 \%$ petroleum ether and recrystallized from acetone. Crystal approx. $0.20 \times 0.05 \times 1.00 \mathrm{~mm}$. Nonius CAD-4F-11M diffractometer, graphite-monochromated radiation, $\omega / 2 \theta$ scan mode, scan speed $1^{\circ} \min ^{-1}$,


Fig. 1. A perspective view of the molecule with atomic numbering. © 1988 International Union of Crystallography


[^0]:    *Lists of structure factors, anisotropic thermal parameters and
    H-atom parameters have been deposited with the British Library
    Document Supply Centre as Supplementary Publication No. SUP
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